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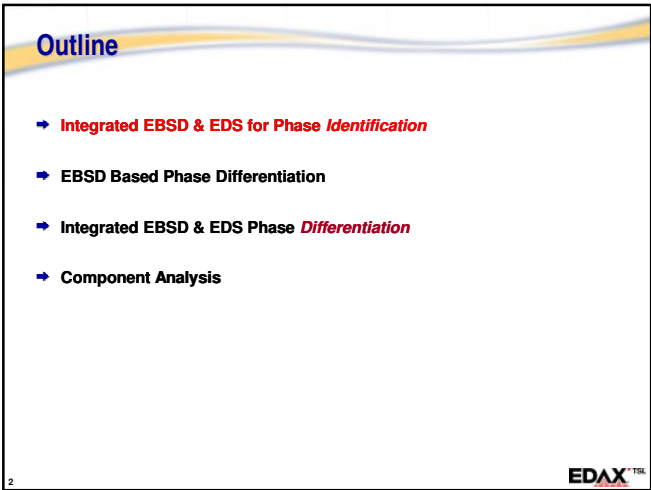
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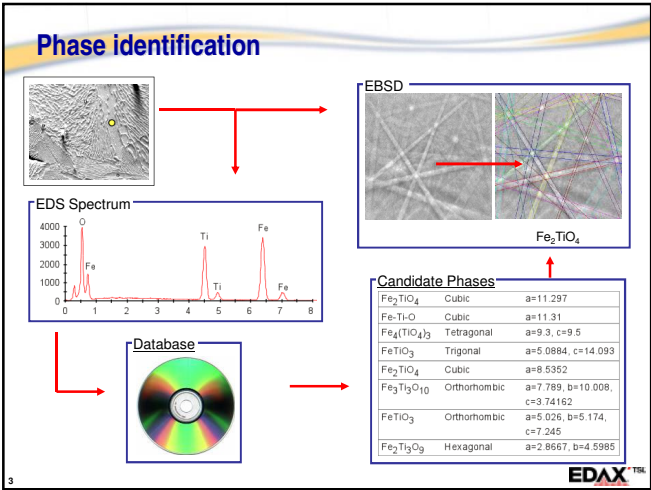
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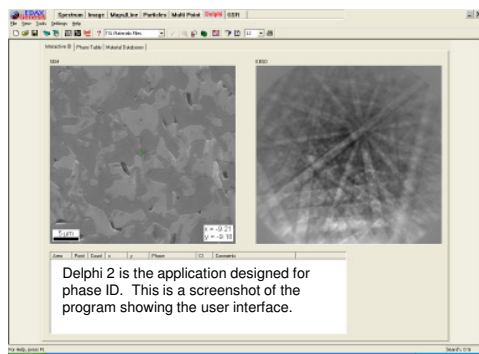
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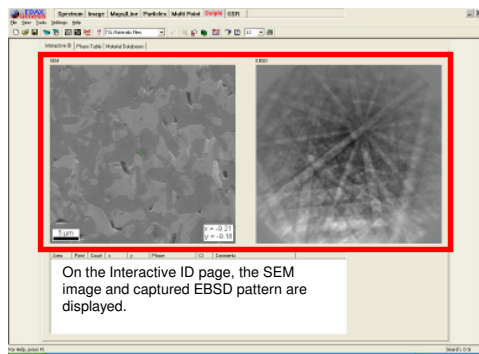
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## Delphi 2



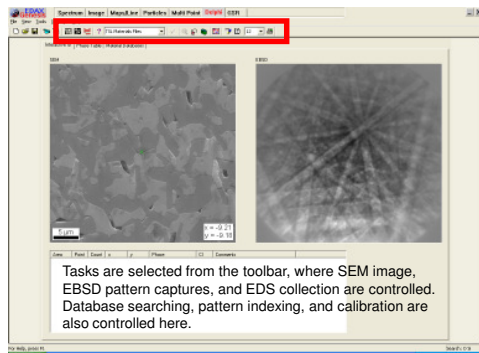
EDAX™ TSL

## Delphi 2



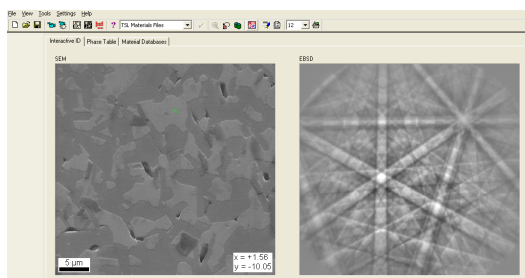
EDAX™ TSL

## Delphi 2



EDAX™ TSL

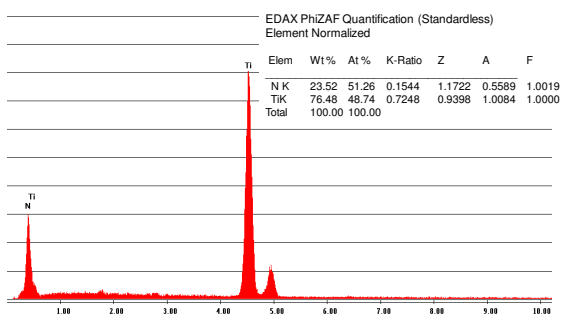
## Delphi 2



An analysis location is selected in the SEM image, and an EBSD pattern collected from that position.

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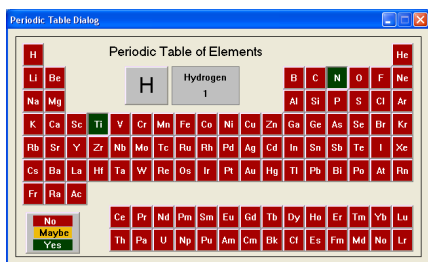
## Example 1



An EDS spectrum is also collected. Here Ti and N were identified as being present. Quant analysis indicates 51% N and 49% Ti

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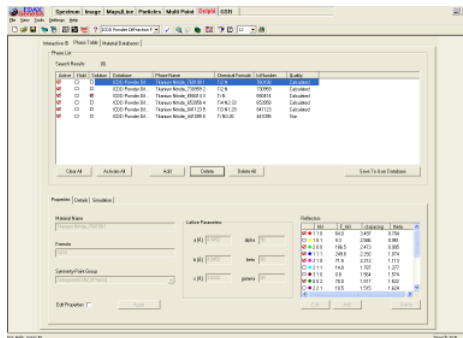
## Example 1



The periodic table is used to specify which elements to search for in the different databases. A Yes/No/Maybe search for each element is possible. Here Ti and N are considered to be present.

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## Example 1



After duplication filtering, 6 candidates with this elemental chemistry are found. The results are shown on the Phase Table page.

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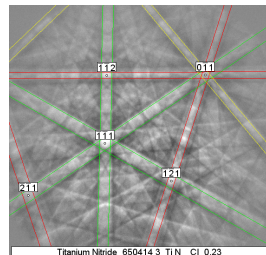
## Example 1

The phase differentiation procedure is then used with these 6 candidate phases.

The TiN phase (PDF#6504143) initially has the highest Rank Factor, as well as the highest CI value. It also has the lowest fit value.

However, it has only the 2nd highest number of votes (20 vs. 30).

Visual inspection of the solution shows while the displayed bands match, only some of the observed bands are accounted for.

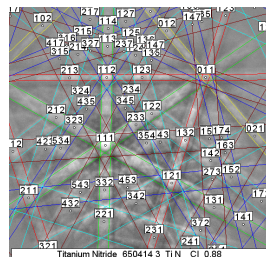


Phase	Votes	Fit (%)	CI	Rank Factor
<input checked="" type="checkbox"/> Titanium Nitride_650414.3	20	0.60	0.226	3.00
<input type="checkbox"/> Titanium Nitride_653958.4	5	1.17	0.012	1.07
<input type="checkbox"/> Titanium Nitride_730959.2	30	2.42	0.060	0.76
<input type="checkbox"/> Titanium Nitride_441095.6	4	2.05	0.000	0.45
<input type="checkbox"/> Titanium Nitride_760198.1	4	3.51	0.000	0.34
<input type="checkbox"/> Titanium Nitride_841123.5	4	5.56	0.000	0.21

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## Example 1

Reflectors	hkl	F <sub>hkl</sub>	d-spacing	theta
<input checked="" type="checkbox"/> 111	177.5	2.460	1.000	
<input checked="" type="checkbox"/> 200	333.0	2.130	1.195	
<input checked="" type="checkbox"/> 220	76.7	1.506	1.634	
<input checked="" type="checkbox"/> 311	14.8	1.284	1.916	
<input checked="" type="checkbox"/> 331	4.8	0.977	2.519	
<input checked="" type="checkbox"/> 420	9.9	0.953	2.584	
<input checked="" type="checkbox"/> 422	8.1	0.870	2.831	
<input checked="" type="checkbox"/> 511	4.4	0.820	3.003	



A better match is found by manually inspecting bands and activating reflectors initially deactivated due to intensity.

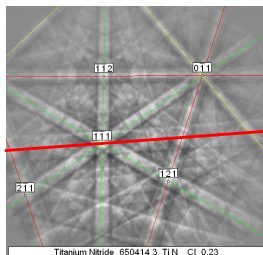
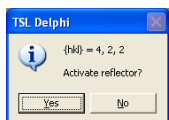
Phase	Votes	Fit (%)	CI	Rank Factor
<input checked="" type="checkbox"/> Titanium Nitride_650414.3	82	0.59	0.881	3.00
<input type="checkbox"/> Titanium Nitride_653958.4	5	1.17	0.012	1.02
<input type="checkbox"/> Titanium Nitride_730959.2	30	2.42	0.060	0.56
<input type="checkbox"/> Titanium Nitride_441095.6	4	2.05	0.000	0.45
<input type="checkbox"/> Titanium Nitride_760198.1	4	3.51	0.000	0.34
<input type="checkbox"/> Titanium Nitride_841123.5	4	5.56	0.000	0.21

EDAX™



## Example 1

Reflector	hkl	F <sup>2</sup> hkl	d-spacing	theta
<input checked="" type="checkbox"/> 111	111	177.5	2.460	1.000
<input checked="" type="checkbox"/> 200	200	333.0	2.130	1.195
<input checked="" type="checkbox"/> 220	220	76.7	1.506	1.634
<input checked="" type="checkbox"/> 311	311	14.8	1.204	1.916
<input checked="" type="checkbox"/> 331	331	4.8	0.977	2.519
<input checked="" type="checkbox"/> 420	420	9.9	0.953	2.584
<input checked="" type="checkbox"/> 422	422	8.1	0.870	2.831
<input checked="" type="checkbox"/> 511	44	0.820	3.003	

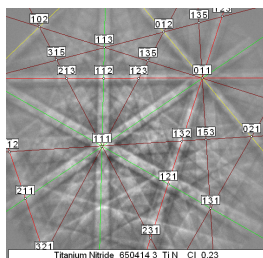


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## Example 1

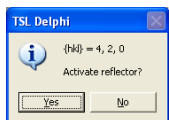
When manually activating or adding reflectors, equivalent planes are also used. This can be used as a test to see if the correct structure is being used.



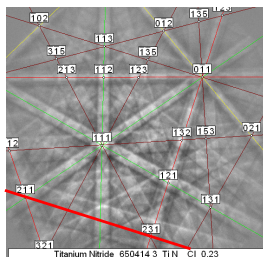
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## Example 1



hkl	F <sup>2</sup> hkl	d-spacing	theta
<input checked="" type="checkbox"/> 111	177.5	2.460	1.000
<input checked="" type="checkbox"/> 200	333.0	2.130	1.195
<input checked="" type="checkbox"/> 220	76.7	1.506	1.634
<input checked="" type="checkbox"/> 311	14.8	1.204	1.916
<input checked="" type="checkbox"/> 331	4.8	0.977	2.519
<input checked="" type="checkbox"/> 420	9.9	0.953	2.584
<input checked="" type="checkbox"/> 422	8.1	0.870	2.831
<input checked="" type="checkbox"/> 511	4.4	0.820	3.003



Determines that this band is the {420} band.

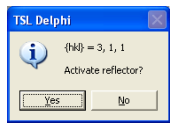
Option to Activate this Band into the Indexing routine

Checks the database file to see if this {420} band is present.

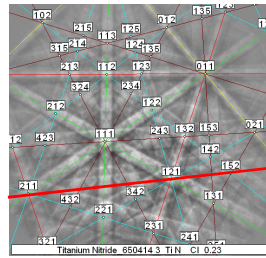
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EDAX<sup>TM</sup>

## Example 1



	hkl	F_hkl	dspacing	theta
✓	111	1775	2.460	1.000
✓	200	333.0	2.130	1.195
✓	220	76.7	1.536	1.534
✓	311	14.8	1.284	1.916
✓	331	4.8	0.977	2.519
✓	420	5.9	0.953	2.584
✓	422	6.1	0.870	2.891
✓	511	4.4	0.820	3.003



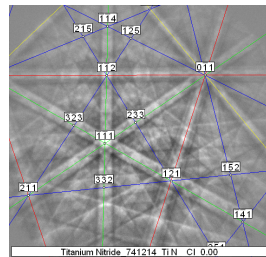
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## Example 1

	hkl	F_hkl	dspacing	theta
✓	111	1745	2.540	0.969
✓	200	333.0	2.200	1.119
✓	220	77.3	1.556	1.582
✓	311	14.8	1.327	1.895

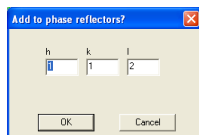
For some database files, the list of possible reflectors is incomplete.



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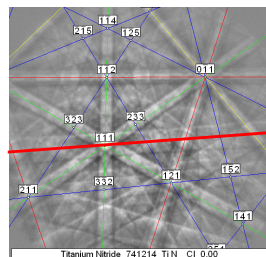
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## Example 1



Instead of activating expected reflectors, manual inspection will add reflectors to the material file.

In this example, a low index ( $h+k+l$ =low number) plane is identified, which is what is expected for EBSD.

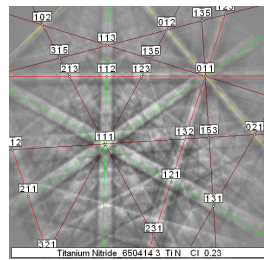


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## Example 1

Just as when activating reflector, when adding reflectors, if the symmetrically equivalent bands also match in the pattern then this is a good confirmation of the correct phase identification.



Titanium Nitride 650414.3 TiN Cl 0.23

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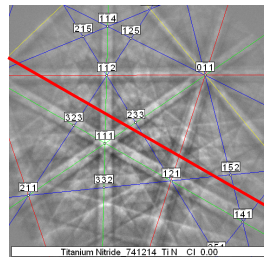
## Example 1

Add to phase reflectors?

h	k	l
10	4	11

OK Cancel

For an arbitrary line however, a high index plane is identified. Care and judgment must be used when adding a reflector.



Titanium Nitride 741214 TiN Cl 0.00

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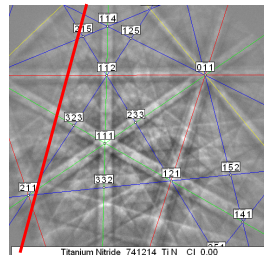
## Example 1

Add to phase reflectors?

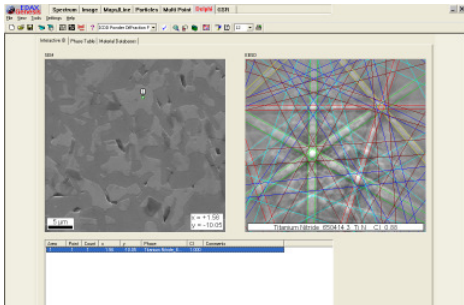
h	k	l
10	23	1

OK Cancel

Also watch for rounding errors when manually drawing lines. The plane should be a [1-20] plane.




# Example 1



The screenshot displays the ImageJ software window titled "Spectrum: ImageJ\_Maps2\_Int\_PlotArea\_MultiPhase [ImageJ].ind". The main workspace is divided into two panels. The left panel shows a grayscale phase-contrast image of a material with a scale bar of 5.0 μm. The right panel shows a corresponding phase map with a complex, multi-colored pattern. Below the images, a status bar indicates the current image is "Image [0] of 1" and the current phase is "Phase 0".

The phase solution is then recorded for this analysis point. The phase, with reflector adjustment, can then be saved to a user database for OIM mapping and additional phase ID.

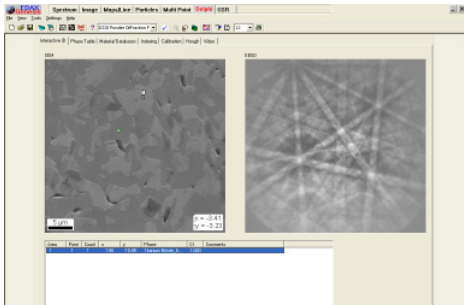


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# Example 1



The screenshot displays the EDAX software interface. The top window shows a scanning electron micrograph (SEM) of a sample. The bottom window shows an energy-dispersive X-ray (EDS) spectrum. The spectrum has a title bar that reads "EDS" and a plot area with a y-axis labeled "Counts" and an x-axis labeled "Energy (keV)". The plot shows a series of peaks, with the most prominent ones labeled "Ti" and "B". Below the plot, a table lists the detected elements and their corresponding atomic numbers and weights.

Element	Atomic Number	Weight
Ti	22	47.88
B	5	10.81

The darker "phase" is then selected in the SEM image, and a pattern obtained. EDS analysis indicated Ti and B are present.

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TECH

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## Example 1

Phase	Votes	F <sub>T</sub>	C <sub>I</sub>	dSpace F <sub>T</sub>	Rank F <sub>c</sub>
<input checked="" type="checkbox"/> Boron Tfx.	94	0.20	0.97	1.00	2.00
<input type="checkbox"/> Boron Tfx.	35	1.72	0.28	1.00	0.56
<input type="checkbox"/> Titanium B.	4	1.98	0.024	1.00	0.22
<input type="checkbox"/> Boron Tfx.	4	2.25	0.024	1.00	0.20
<input type="checkbox"/> Boron Tfx.	2	2.38	0.012	1.00	0.18
<input type="checkbox"/> Boron Tfx.	6	3.00	0.012	1.00	0.15
<input type="checkbox"/> Titanium B.	4	2.86	0.000	1.00	0.14
<input type="checkbox"/> Boron Tfx.	3	3.19	0.000	1.00	0.13
<input type="checkbox"/> Boron Tfx.	4	3.66	0.000	1.00	0.11
<input type="checkbox"/> Titanium Tfx.	1	3.76	0.000	1.00	0.11
<input type="checkbox"/> Titanium B.	1	5.63	0.000	1.00	0.07
<input type="checkbox"/> Boron Tfx.	4	10.24	0.012	1.00	0.05

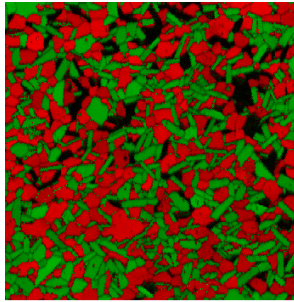
Boron Titanium 651973 S. 82 Ti Cl 0.92

The second phase is identified as  $\text{TiB}_2$  following the same procedure.

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## Example 1



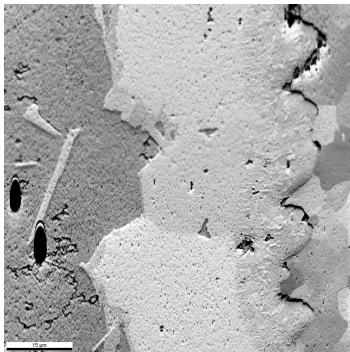
Red: TiN  
Green: TiB<sub>2</sub>

With these material files, it is then possible to obtain a multi-phase OIM map.

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## Example: phase identification



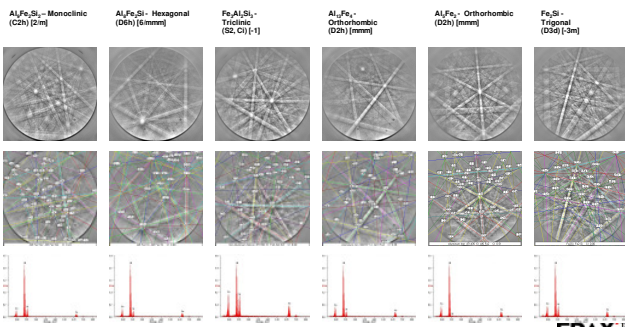
- FSD image of a Al – Fe weld interface.
- At the interface Si was added to facilitate the welding process.
- A large number of phases could be identified on the BSE and FSD (left) images.
- All the phases contain Al, Fe, and / or Si in different quantities and may be part of continuous solid-solution series.

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## Phase identification: Al-Fe (Si) welding interface

- The elements are identified for each phase
- This chemistry is used to search a database of candidate phases (e.g. ICDD database)
- The candidates are matched against the pattern using dedicated Phase ID software, Delphi.

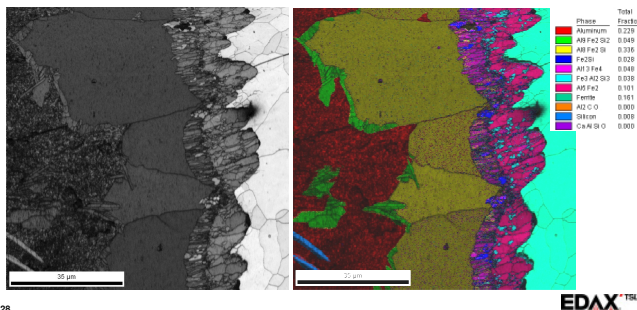


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## EBSD map results

After identification of all phases, an EBSD map was collected across the interface.



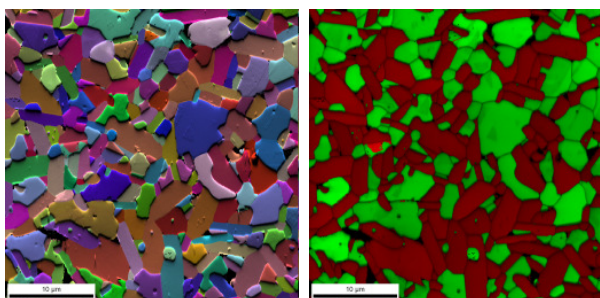
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## Outline

- ➔ Integrated EBSD & EDS for Phase **Identification**
- ➔ EBSD Based Phase **Differentiation**
- ➔ Integrated EBSD & EDS Phase **Differentiation**
- ➔ Component Analysis

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## OIM – multiphase samples TiN - TiB<sub>2</sub> (cub – hex)



- ➔ 195 Indexed Points per Second with 96% Success Rate

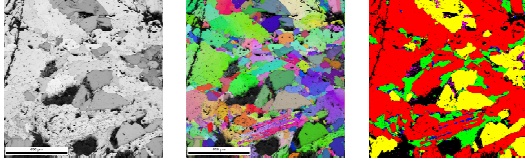


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## Common issues with EBSD on multiphase materials

- Large variation in EBSD pattern quality due to different polishing properties.



- Need to index each pattern with all possible phases and then select the best solution. This is slow and may produce errors.
- Phases may have the same crystal structure and cannot be differentiated from each other by EBSD (e.g. Ni vs. Cu).

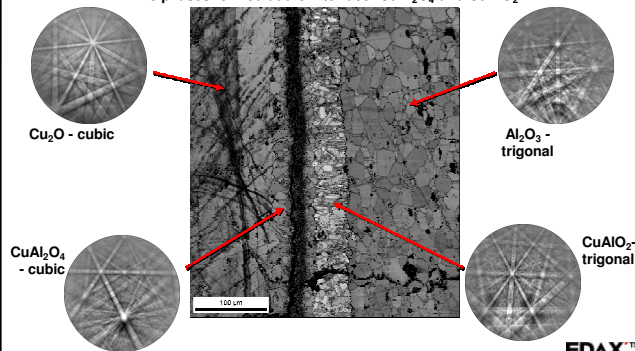
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## Multi-Symmetry Sample

### Copper oxide – Aluminium oxide reaction couple

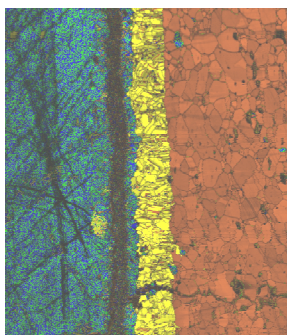
Two phases formed at the interface:  $\text{CuAl}_2\text{O}_4$  and  $\text{CuAlO}_2$



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## Conventional phase differentiation



Both trigonal phases,  $\text{Al}_2\text{O}_3$  (orange) and  $\text{CuAlO}_2$  (yellow) are successfully identified

The cubic phases  $\text{Cu}_2\text{O}$  (blue) and  $\text{CuAl}_2\text{O}_4$  (green) could not be distinguished

Gray Scale Map Type: Image Quality  
0.261432 (0.261432)

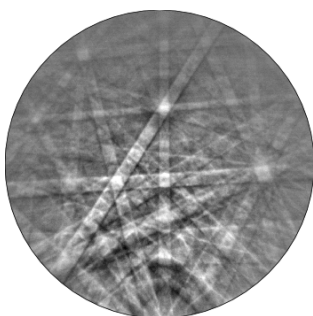
Phase	Fraction	Partition
$\text{Al}_2\text{O}_3$	0.417	0.422
$\text{CuAl}_2\text{O}_4$	0.140	0.142
$\text{CuAlO}_2$	0.120	0.121
$\text{Cu}_2\text{O}$	0.312	0.315

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## Scanning multiphase materials – phase 1

1. Obtain a pattern



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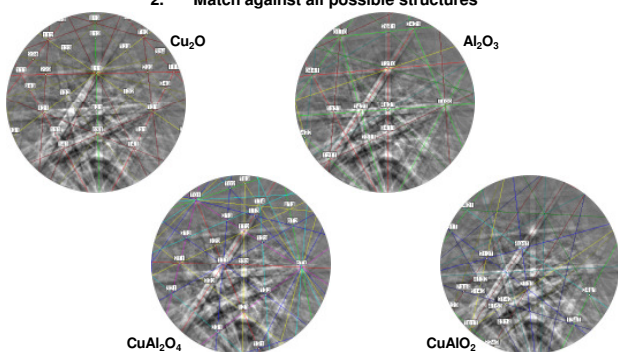
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## Scanning multiphase materials – phase 1

2. Match against all possible structures



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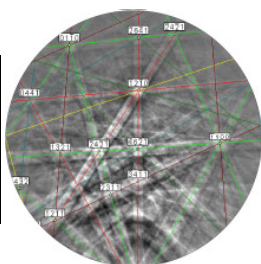
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## Scanning multiphase materials – phase 1

3. Select the best fit solution

phase	votes	fit (°)	confidence index	rank
CuO <sub>2</sub>	6	2.10	0.000	0.48
Al <sub>2</sub> O <sub>3</sub>	80	0.86	0.500	3.00
CuAlO <sub>2</sub>	12	2.10	0.017	0.59
CuAl <sub>2</sub> O <sub>4</sub>	8	2.11	0.008	0.52



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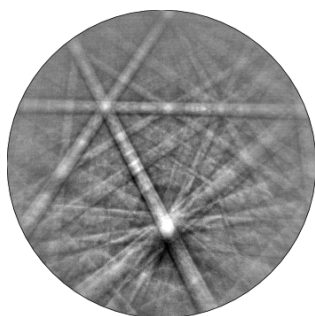
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## Scanning multiphase materials – phase 2

1. Obtain a pattern



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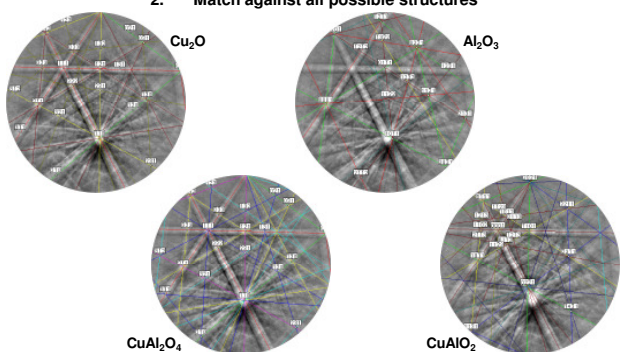
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## Scanning multiphase materials – phase 2

2. Match against all possible structures



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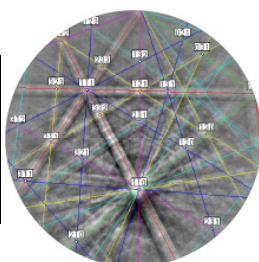
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## Scanning multiphase materials – phase 2

3. Select the best fit solution

Both phases are cubic

phase	votes	fit (°)	confidence index	rank
$\text{Cu}_2\text{O}$	56	0.46	0.821	3.00
$\text{Al}_2\text{O}_3$	10	1.78	0.018	0.46
$\text{CuAlO}_2$	13	1.69	0.107	0.63
$\text{CuAl}_2\text{O}_4$	56	0.46	0.821	3.00



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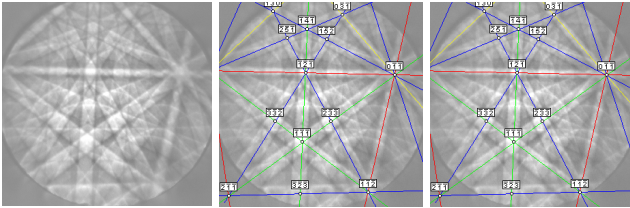
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## Phase differentiation

Nickel Pattern

Indexed as Nickel

Indexed as Copper



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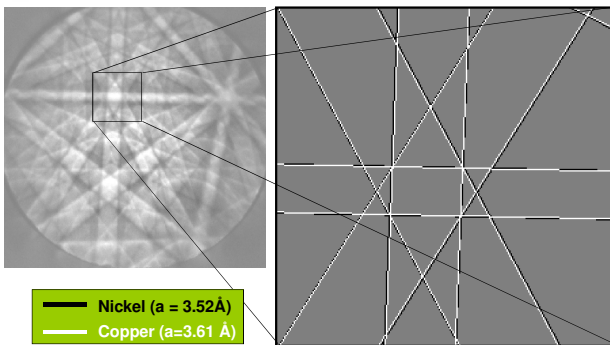
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## Phase differentiation – band widths



1024 x 1024 Image

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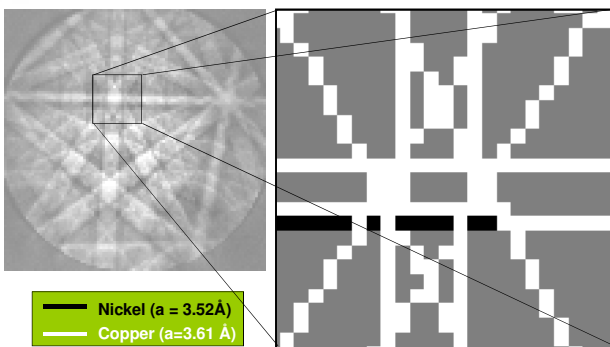
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## Phase differentiation – band widths



96 x 96 Image

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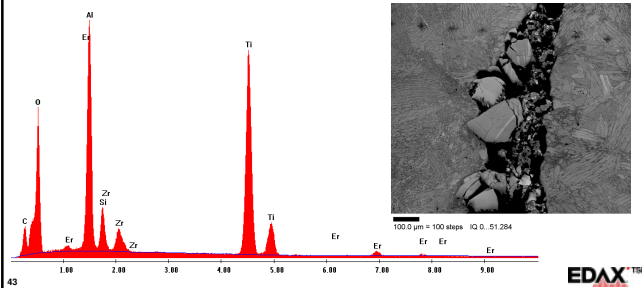
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## Sample

Titanium-aluminum sample with inclusions.

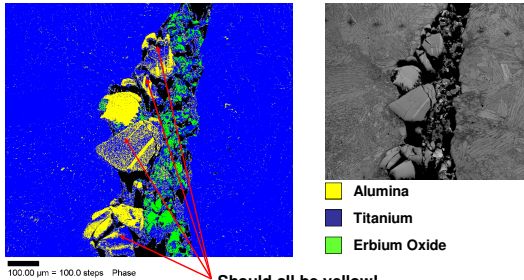
The EDS spectrum showed that titanium, aluminum, oxygen, zirconium and erbium were present in the material.

We knew a-priori that the two main phases were a hexagonal Ti phase and trigonal Alumina. We also identified an Erbium Oxide phase and two Zirconium Oxides.



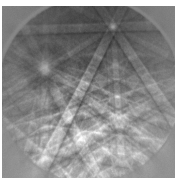
## Conventional phase differentiation

It was obvious from the phase map, that the software had a difficult time differentiating the Titanium and Alumina phases reliably. We tried modifying some of the indexing parameters but with about 870,000 data points these off-line iterations were fairly time consuming (about 3 hours each). Therefore, it was difficult to optimize the settings to improve the results.

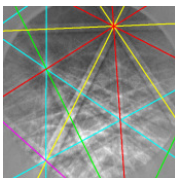


## Conventional phase differentiation

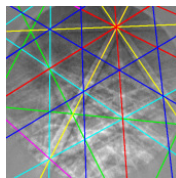
Pattern from Alumina



Indexed as Alumina

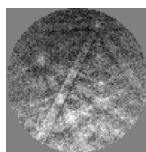


Indexed as Titanium

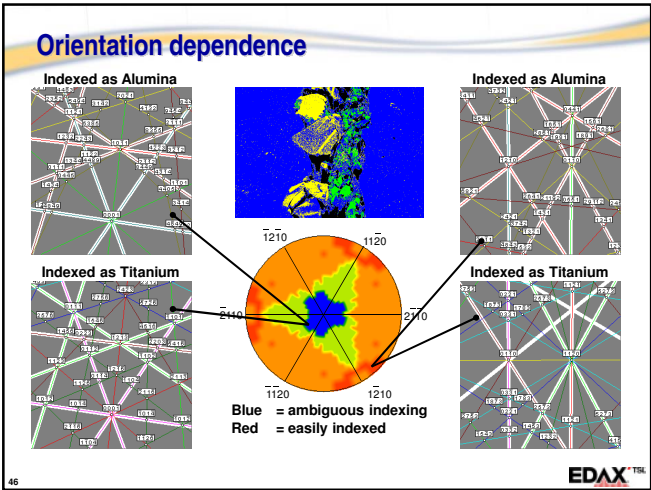


Real full speed scan patterns are much weaker than the beautiful high resolution pattern above.

8x8 binning & then compressed to 96x96



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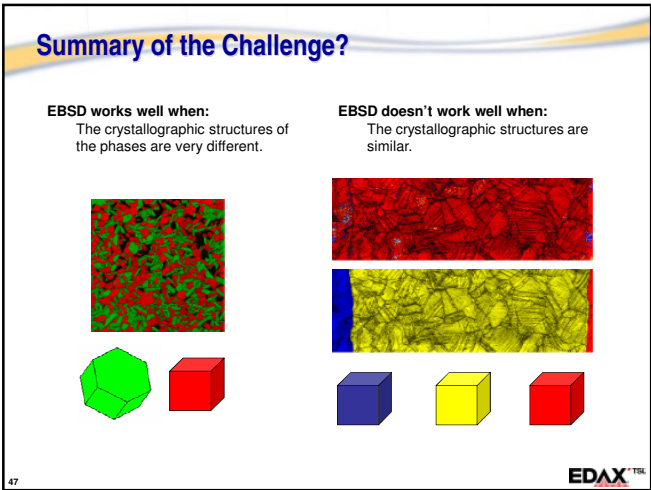
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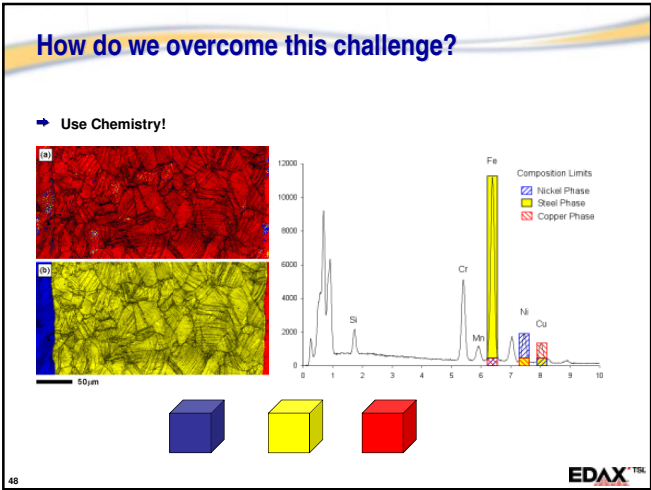
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## Outline

- Integrated EBSD & EDS for Phase **Identification**
- EBSD Based Phase Differentiation
- **Integrated EBSD & EDS Phase Differentiation**
- Component Analysis

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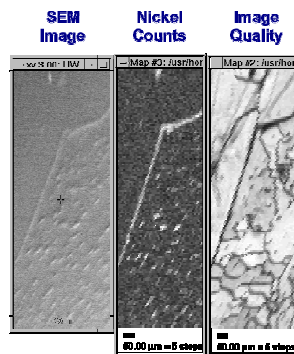
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## Combined EBSD + EDS

- Simultaneous EDS and EBSD mapping have been available for 14 years.



S. I. Wright (1997) Unpublished work at the University of Wisconsin-Madison. Presented at Microscopy and Microanalysis, 1997 Cleveland, Ohio.

50

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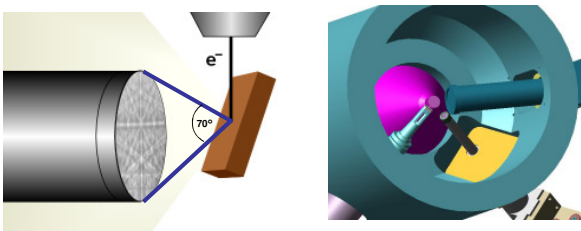
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## Detector considerations

- For good simultaneous EDS - EBSD analysis, the sample surface has to be within the line of sight of both detectors.
- The EBSD detector must see a proper solid angle.
- There should be no detector shadowing.



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## Phase differentiation using simultaneous EDS and EBSD

- ➔ Analysis area is scanned and at each point the relevant pattern parameters are stored together with the EDS region-of-interest counts.
- ➔ Positions of phases are determined using X-ray maps.
- ➔ During off-line indexing, the recorded chemistry determines which phase / crystal structure file is used for indexing of each point
- ➔ Each pattern is indexed by only one phase

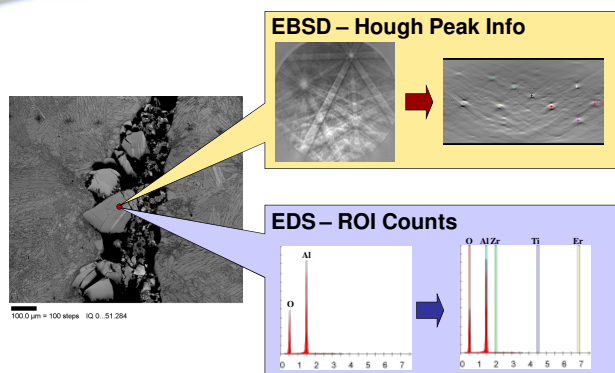
### ChI scan – Chemistry assisted Indexing

M. M. Nowell and S. I. Wright (2004). "Phase differentiation via combined EBSD and XEDS." *Journal of Microscopy* 213: 296-305

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## Data collected

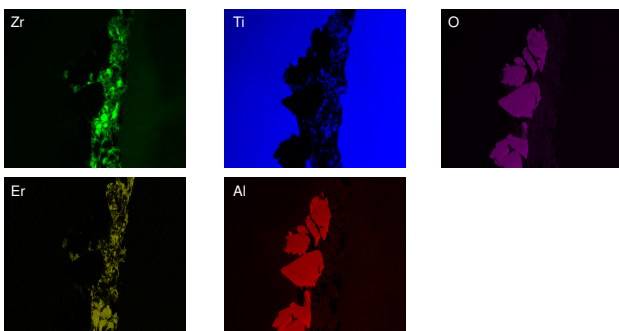


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## Elemental maps

We were able to construct element maps from the EDS data collected simultaneously. This aided us in locating and identifying other phases. We were able to identify and index two more phases: a monoclinic  $\text{ZrO}_2$  phase and a tetragonal  $\text{ZrO}_2$  phase.



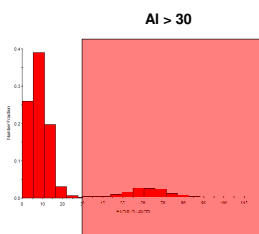
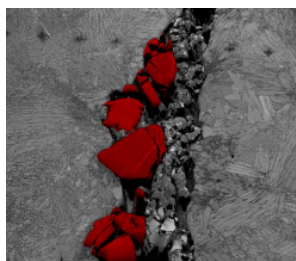
54

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## Chemical filtering ( $\text{Al}_2\text{O}_3$ )

All of the red grains will be indexed as Alumina

Al elemental map



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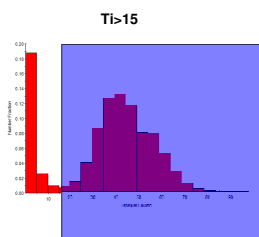
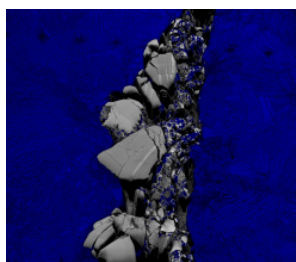
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## Chemical filtering (Ti)

Everything highlighted in blue will be indexed as titanium

Ti elemental map



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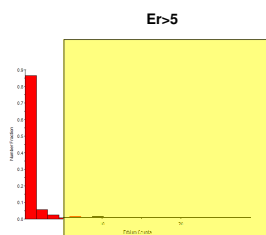
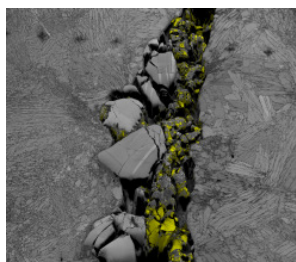
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## Chemical filtering ( $\text{Er}_2\text{O}_3$ )

Everything highlighted in yellow will be indexed as Erbium Oxide.

Er elemental map



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## Chemical filtering ( $\text{ZrO}_2$ )

Everything in green will indexed as either tetragonal Zirconium Oxide or monoclinic Zirconium Oxide. EBSD will be used to differentiate between the two.

Zr elemental map

The figure displays two EBSD maps of a zirconium oxide sample. The top map is a color-coded orientation map showing various shades of blue, green, and yellow, representing different crystallographic orientations. The bottom map is a Zr elemental map, where the sample area is predominantly green, indicating the presence of zirconium. A scale bar in the bottom left corner indicates a length of 100.0 μm. The bottom right corner contains technical data: IQ 3.253...209.584, EDS [Zirconium] 10...66.

100.0 μm ≈ 100 steps IQ 3.253...209.584, EDS [Zirconium] 10...66

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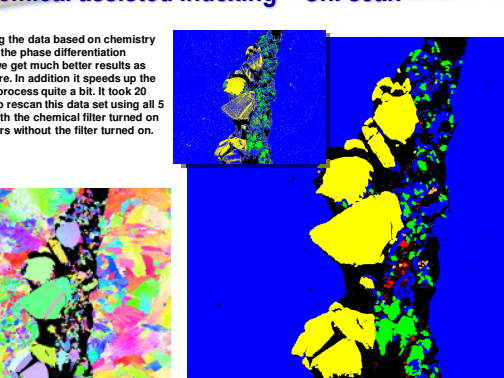
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## Chemical assisted indexing – Chl scan

By filtering the data based on chemistry as part of the phase differentiation process we get much better results as shown here. In addition it speeds up the indexing process quite a bit. It took 20 minutes to rescan this data set using all 5 phases with the chemical filter turned on and 3 hours without the filter turned on.



100.0  $\mu\text{m}$  = 100 steps Phase

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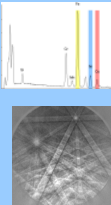
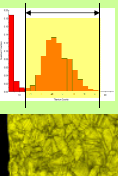
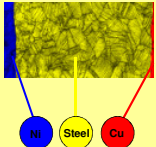

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# Flowchart

<p>Collect EDS &amp; EBSD data</p>  <p>Blue arrow pointing right</p>	<p>Set up an elemental filter for each phase present</p> <p>Steel</p>  <p>Green arrow pointing right</p>	<p>Assign each scan point a crystallographic phase(s) based on the chemical filter</p>  <p>Yellow arrow pointing right</p>	<p>Index EBSP's at each scan point using the phase information assigned to each scan point</p>  <p>Red curved arrow pointing left</p>
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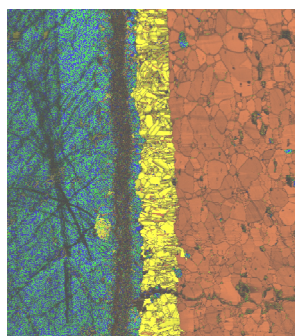
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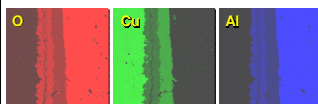


## ChI scan phase differentiation – Example 2



Both trigonal phases,  $\text{Al}_2\text{O}_3$  (orange) and  $\text{CuAlO}_2$  (yellow) are successfully identified

The cubic phases  $\text{Cu}_2\text{O}$  (blue) and  $\text{CuAl}_2\text{O}_4$  (green) could not be distinguished with EBSD alone, but are clear in the X-Ray maps (below)

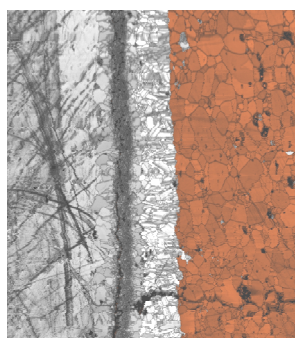


Phase  
 $\text{Al}_2\text{O}_3$   
 $\text{CuAl}_2\text{O}_4$   
 $\text{CuAlO}_2$   
 $\text{Cu}_2\text{O}$

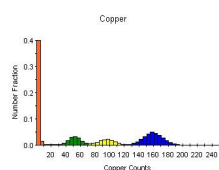
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## Chemical filtering



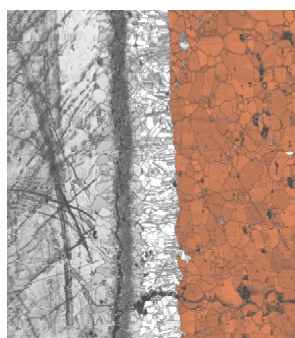
$\text{Cu}_2\text{O}$   
 $\text{CuAlO}_2$   
 $\text{CuAl}_2\text{O}_4$   
 $\text{Al}_2\text{O}_3$



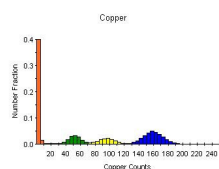
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## Chemical filtering



These phase definitions can now be used to select the proper crystal structure for each pixel in the scan, and distinguish crystallographically similar phases.



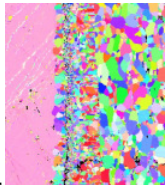
EDAX™ TSL

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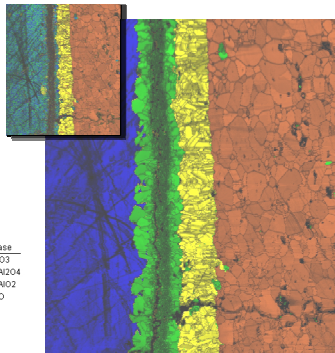
## Chi-Scan results

By filtering the data based on chemistry as part of the phase differentiation process it is now possible to differentiate between the cubic phases and the microstructural properties of all phases may be investigated.

In addition, indexing speed is greatly improved from 1 hour (conventional scanning) to 10 minutes when rescanning this dataset using all 4 phases with the chemical filter turned on.



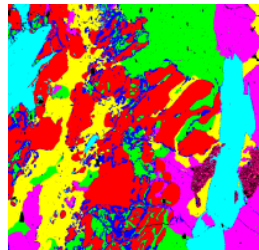
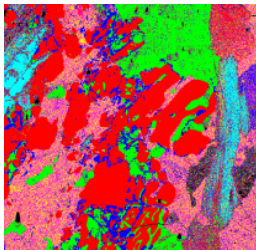
Phase  
Al2O3  
CuAl2O4  
CuAlO2  
CuO



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## Example - mineral



Phases	Chemical Limits (EDS Counts)
Nickeline	Ni>10, As>45, S<30
Pyrrhotite	Ni<10, S>20, As<10, Cu<10, Fe>40
Chalcopyrite	S>20, Ni<10, As<10, Cu>10, Fe>15
Cobaltite	As>10, S>15
Pentlandite	S>20, Fe>15, Cu<10, Ni>10
Biotite	K>10
Carbon	C>20

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## Outline

- ➔ Integrated EBSD & EDS for Phase **Identification**
- ➔ EBSD Based Phase Differentiation
- ➔ Integrated EBSD & EDS Phase **Differentiation**
- ➔ **Component Analysis**

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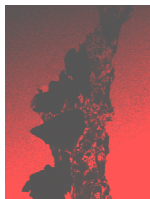
66

## Automated phase recognition (PCA)

### Issues with phase selection based on chemistry:

- Variation in EDS intensity over the scan area
  - 1- Because of the high-tilts required for EBSD, there is often a change in the EDS signal with WD
  - 2- Beam instabilities may cause variations in counts rates during long scans
- Difference in spatial resolution of EDS and EBSD

The spatial resolutions of the two techniques are approximately 50 nm and 1 micron for EBSD and EDS respectively. Thus, there will be some "smearing" at the boundaries where the EBSD must be used exclusively for the phase differentiation.



These issues can be minimised with Automated Phase Recognition

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## Phase cluster analysis

### A method of statistical analysis of chemistry data:

- ➔ Groups pixels based on similarity in the chemistry (EDS ROI counts).
- ➔ Allows the user to automatically find phases in the recorded data without prior knowledge.
- ➔ PCA ChI scan bridges the gap between the spatial resolutions of EBSD and EDS.
  - Grains down to 200 nm can now successfully be defined.

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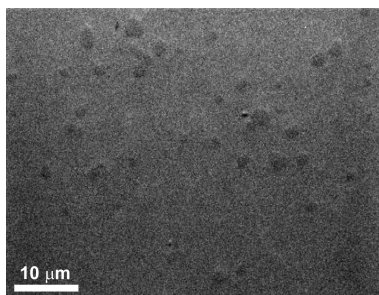
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## Silicon and Chromium Carbides in Cobalt Matrix



SE image

69

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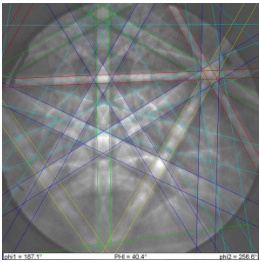
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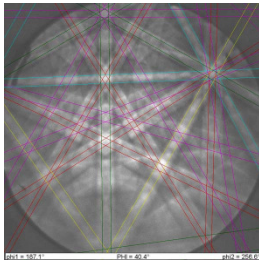
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EBSD Patterns



Indexed as Co



Indexed as CrC

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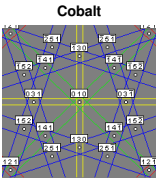
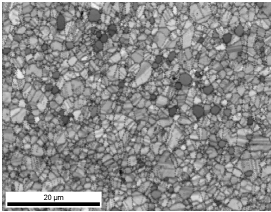
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PCA example

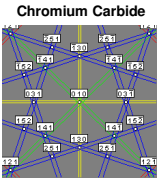
A three phase sample:

- 1) Cobalt
- 2) Chromium Carbide
- 3) Silicon Carbide

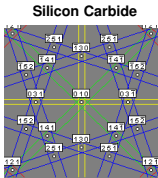
All three phases posses fcc cubic crystal symmetry making it very difficult to distinguish them from each other using EBSD alone.



Cobalt



Chromium Carbide



Silicon Carbide

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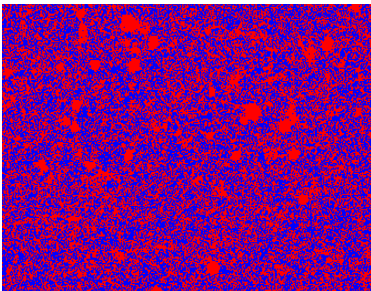
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Indexing with crystallography only



14.00 µm ~ 70 steps Phase

Phase  
Cobalt  
Chromium Carbide

72

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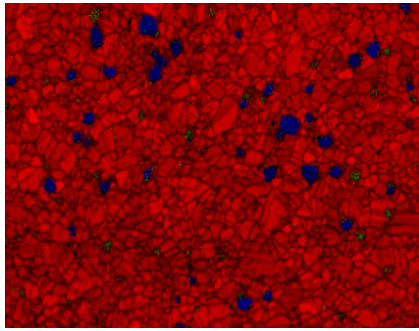
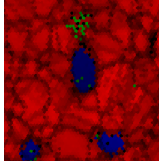
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## Results from the manually set limits

- Cobalt
- Chromium Carbide
- Silicon Carbide

Difficult to find good sets of limits because of limited resolution.



20 μm

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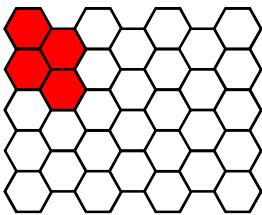
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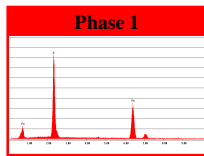
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## Phase cluster analysis



Step 1 - The spectra from the first 4 pixel block of measurement points are taken together and are assumed to be from a single phase.

The EDS signal from these pixels is defined as reference phase 1.



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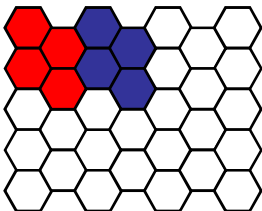
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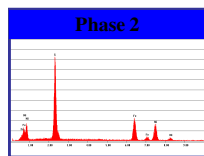
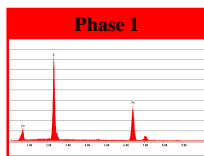
## Phase cluster analysis



Step 2-The average spectrum from the second 4x4 block is compared to the first component. (Using a "contingency coefficient" based on the sum of squared differences calculated from normalized spectra)

$$\chi^2 = \sum_{i=1}^N \frac{(S_i - s_i)^2}{S_i} \quad C = \frac{\chi^2}{\chi^2 + N}$$

The "spectra" are normalized before calculating the  $\chi^2$  values. N is the number of channels in the EDS spectra.  $S_i$  and  $s_i$  are the number of counts in the  $i^{th}$  channel of spectra S and s. C is the contingency coefficient.



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Reference: Numerical recipes in C: The art of scientific computing, E. H. Press, S. A. Teukolsky, W. T. Vetterling and B. P. Flannery, Cambridge University Press, Cambridge (1995).

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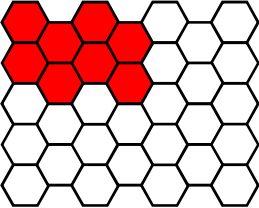
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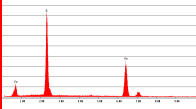
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**Phase cluster analysis**

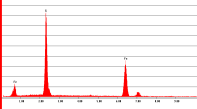


Step 3 - Step 3-If the second block matches the first component within a given tolerance of the contingency coefficient, this second spectrum is added to the first component.

Phase 1



Phase 1



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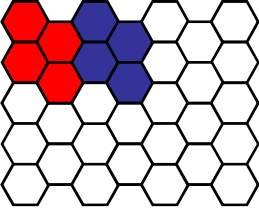
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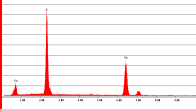
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**Phase cluster analysis**

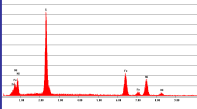


Step 4 - If the second block does not match the reference chemistry, it then defines a reference phase 2.

Phase 1



Phase 2



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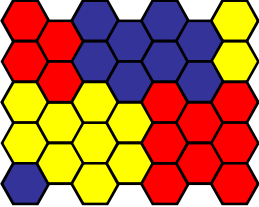
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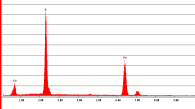
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**Phase cluster analysis**

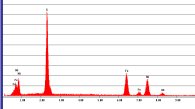


Step 5 - This comparative process is continued until each 4 pixel block has been matched to a phase.

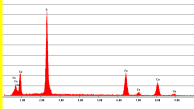
Phase 1



Phase 2



Phase 3



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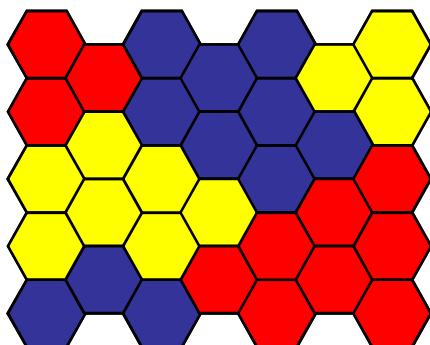
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## Phase cluster analysis



Step 6 –  
When all reference  
phases are defined,  
the chemistry of each  
individual pixel is  
compared with and  
matched to one of the  
determined phases.

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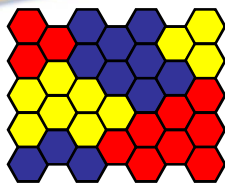
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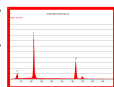
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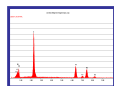
## Phase cluster analysis



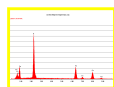
Step 7 - Candidate phases are  
assigned to each chemical  
component.



Phase A



Phase B  
Phase C



Phase D

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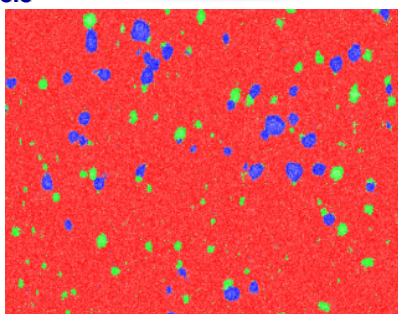
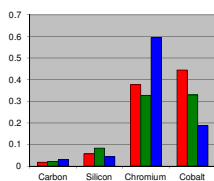
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## Component analysis

The component analysis  
found three components.  
With the following  
"spectra". The variation in  
the hue shows the  
difference of the individual  
pixel from the average for  
the component.



Component 1 → Cobalt  
Component 2 → Chromium Carbide  
Component 3 → Silicon Carbide

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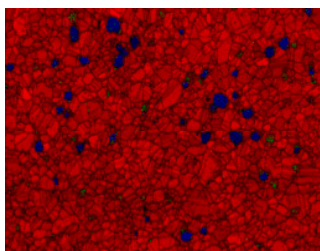
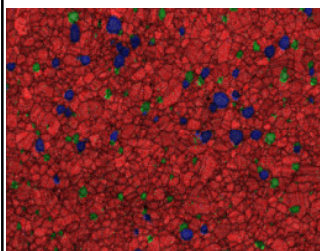
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## Component analysis - comparison

Component Analysis

Manual Limit Setting

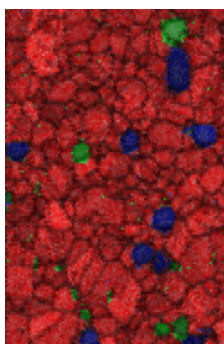


■ Cobalt
 ■ Chromium Carbide
 ■ Silicon Carbide

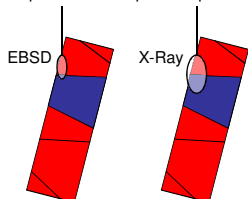
82

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## Comparison of PCA with Manual Limits



Note the "bleed" below some of the blue and green grains. This is a resolution difference effect. The bottom of the map corresponds to the top of sample.

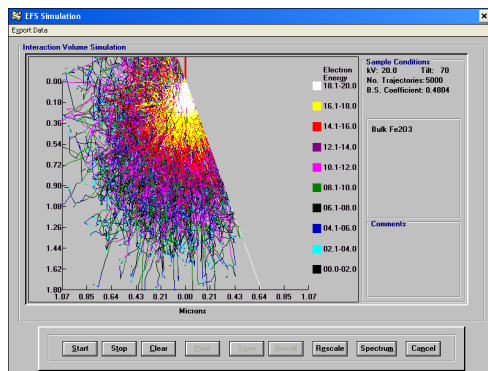


Gray scale in the map is for EBSD, colors are for X-Ray

83

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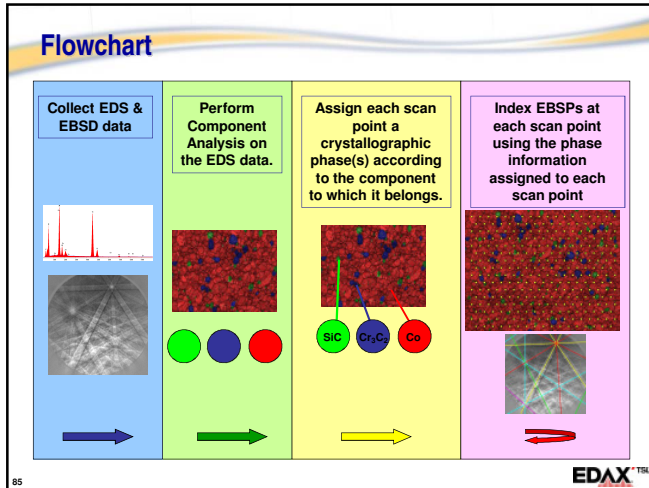
## EDS interaction Volume



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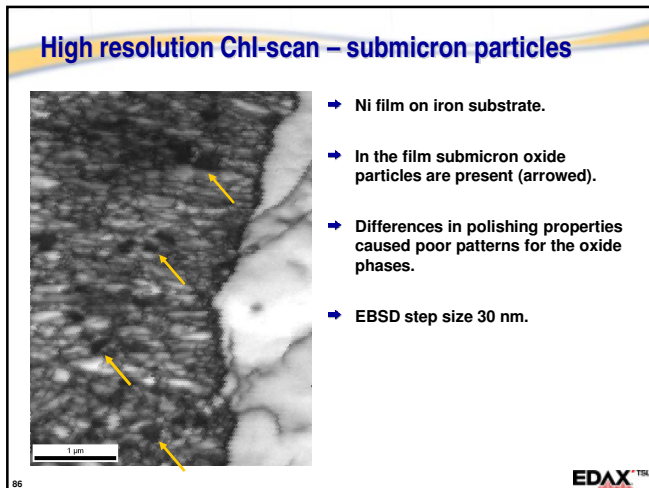
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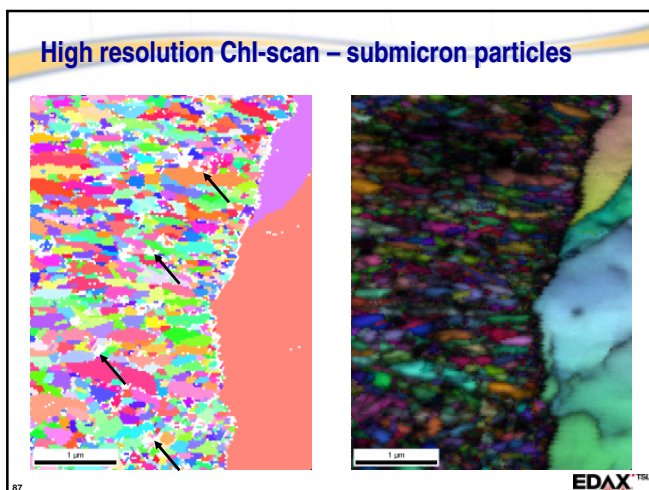
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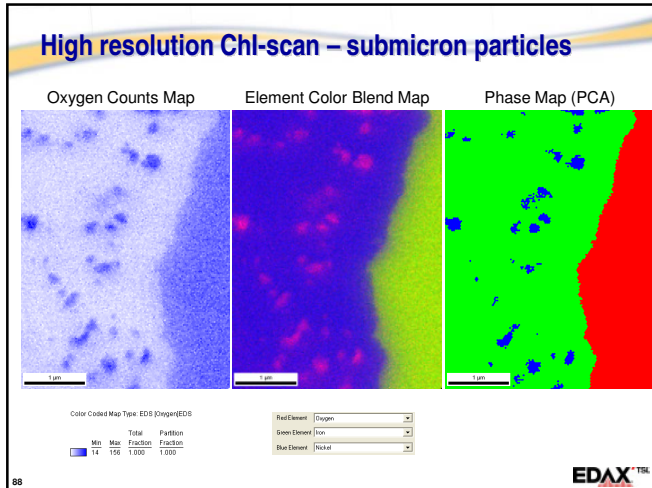
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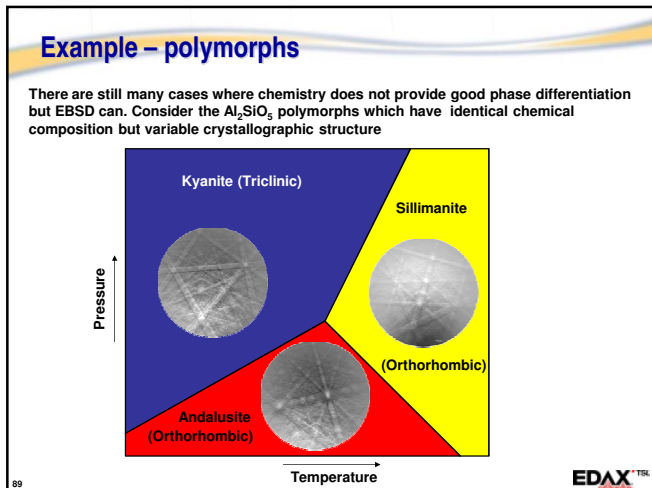
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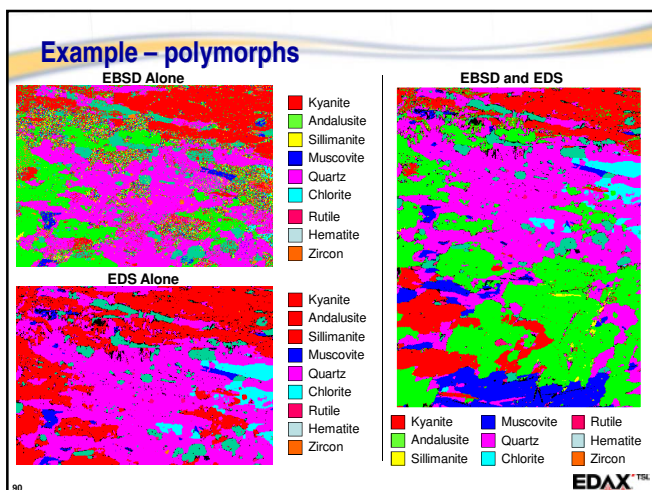
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## Conclusions

Multi-phase indexing with (PCA) ChI-scan enables:

- Distinction of phases with similar crystal structure
- Distinction of phases with similar chemistry
- Greatly improved indexing accuracy
- Allows fast scanning of polyphase materials regardless of the number of phases present
- Minimises effects of different spatial resolutions of EDS and EBSD and variation in EDS intensities
- Typical ChI-scan collection speeds are between 10 and 100 points per second.  
(100 to 10 msec EDS dwell time)

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